# Master Materials and Equipment List

*Italicized entries* indicate items that are not available from PASCO. The <u>For each student group</u> quantity indicated is per student group, except when an asterisk \* is present to indicate a per-class quantity. A double asterisk \* indicates a variable quantity. The <u>For teacher preparation</u> quantity is the amount required per class. Teacher preparation quantities are designed for approximately 8 student groups per class. The volumes or quantities may be adjusted according to a teacher's discretion.

**NOTE:** The activities also require protective gear for each student (for example, safety goggles, gloves, apron, or *lab coat*).

The equipment list is the same whether PASCO Capstone or SPARKvue software is used to conduct the investigation. Spectrometer versions of investigations require Spectrometry software, which is available as a free downland on **pasco.com** for Windows and Mac computers, and is available as a free app in most mobile app stores. Teachers can conduct some lab activities with sensors and probes other than those listed here. For assistance with substituting compatible software, sensors, and probes for a lab activity, contact PASCO Technical Support (800-772-8700 inside the United States or **pasco.com/support**).

Lab #	Title	Materials and Equipment	Qty
1	Analyzing Food Dyes in Sports Drinks	For each student group:	
		Computer or mobile device	1
		PASCO Wireless Colorimeter (PS-3215)	8
		Cuvettes (included with Colorimeter, or SE-8739)	9
		Graduated cylinders, 10-mL	2
		Test tubes, 2-cm x 15-cm	6
		Test tube rack	1
		Disposable pipettes	5
		Sports drink with FD&C Blue #1 food dye	4 mL
		Permanent marker or grease pencil	1
		Lint- and scratch-free lens wipes	2+
		Wash bottle filled with distilled water	1
		Additional drink samples for dye analysis	4 mL per
			sample
		For teacher preparation:	
		FD&C Blue # 1 food dye stock solution	1 L
		FD&C Red #40 food dye stock solution	1 L
		FD&C Yellow #5 food dye stock solution	1 L

Lab #	Title	Materials and Equipment	Qty
2A	Investigating the Copper Content of Brass (Colorimeter)	For each student group:Computer or mobile devicePASCO Wireless Colorimeter (PS-3215)Cuvettes (10 included with Colorimeter; SE-8739)Precision balance (readability: 0.001 g; SE-8860)Beakers, 150-mLVolumetric flask, 100-mLFunnel to fit flaskGraduated cylinders, 10-mLGlass stirring rodWatch glassTest tubes, 15-mL capacity or greaterTest tubes, 0.3 to 1 g eachPermanent marker or grease pencilLint- and scratch-free lens wipesWash bottle filled with distilled waterFor teacher preparation:0.40 M Copper(II) nitrate, Cu(NO_3)_21.0 M Iron(III) nitrate, Fe(NO_3)_31.0 M Nickel nitrate, Ni(NO_3)_21.0 M Nickel nitrate, Ni(NO_3)_21.0 M Nickel nitrate, Ni(NO_3)_2	1 8 13 2* 2 1 1 3 1 7 1 3 * 2 1 2+ 1 500 mL 100 mL 100 mL 100 mL 100 mL
2B	Investigating the Copper Content of Brass (Spectrometer)	For each student group: Computer or mobile device with Spectrometry PASCO Wireless Spectrometer (PS-2600) Cuvettes (10 included with Spectrometer; SE-8739) Precision balance (readability: 0.001 g; SE-8860) Beakers, 150-mL Volumetric flask, 100-mL Funnel to fit flask Graduated cylinders, 10-mL Glass stirring rod Watch glass Test tubes, 15-mL capacity or greater Test tube rack Disposable pipets 6 M Concentrated nitric acid, HNO <sub>3</sub> Brass samples, ~0.3 to 1 g each Permanent marker or grease pencil Lint- and scratch-free lens wipes Wash bottle filled with distilled water For teacher preparation: 0.24 M Copper(II) nitrate, Cu(NO <sub>3</sub> ) <sub>2</sub> 1.0 M Iron(III) nitrate, Fe(NO <sub>3</sub> ) <sub>3</sub> 1.0 M Nickel nitrate, Ni(NO <sub>3</sub> ) <sub>2</sub> 1.0 M Nickel nitrate, Zn(NO <sub>3</sub> ) <sub>2</sub>	1 8 13 2* 2 1 1 3 1 1 7 1 3 * 2 1 2+ 1 2+ 1 500 mL 100 mL 100 mL 100 mL 100 mL 100 mL

Lab #	Title	Materials and Equipment	Qty
3	How Hard is Your Tap Water?	For each student group: Computer or mobile device PASCO Wireless Conductivity sensor (PS-3210) PASCO Wireless Drop Counter with accessories (PS-3214) Precision balance (readability: 0.001 g; SE-8860)	1 1 1 2*
		Beakers, 100-mL Beakers, 250-mL Graduated cylinder, 10-mL Graduated cylinders, 25-mL	3 2 1 2
		Graduated cylinder, 50-mL Watch glass Stirring rod Funnel	1 1 1 1
		Clamp to fit funnel Ring stand Magnetic stirrer with stir bar (SE-7700) Waste beaker	1 1 1 1 1
		Rinse bottle filled with distilled water Optional: Drying oven For teacher preparation:	1 1*
		0.10 M Sodium carbonate, Na <sub>2</sub> CO <sub>3</sub> 0.50 M Sodium carbonate, Na <sub>2</sub> CO <sub>3</sub> 0.075 M Calcium chloride, CaCl <sub>2</sub> 0.10 M Calcium chloride, CaCl <sub>2</sub> 0.15 M Calcium chloride, CaCl <sub>2</sub> 0.50 M Calcium chloride, CaCl <sub>2</sub> Various hard water samples, concentrated	1 L 1 L 1 L 1 L 1 L 1 L 1 L *
4	How Much Acid is in Your Fruit Juice?	For each student group:Computer or mobile devicePASCO Wireless pH sensor (PS-3204)Electrode support or test tube clamp (PS-3505)Beakers, 150-mLBeaker, 250-mLGraduated cylinder, 10-mLMagnetic stirrer with stir bar (SE-7700)Buret, 50-mLBuret, 50-mLBuret clampRing standFunnel to fit buretFruit juices, varietyWash bottle filled with distilled water	1 1 6 1 1 1 1 1 1 1 1 1 1 1 1 1 1
		For teacher preparation: 0.10 M Sodium hydroxide, NaOH 0.10 M Acetic acid, CH <sub>3</sub> COOH	1 L 1 L

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Lab #	Title	Materials and Equipment	Qty
5	Separating Food Dyes Using	For each student group:	
	Chromatography	Computer or mobile device	1
		PASCO Wireless Colorimeter (PS-3215)	1
		Cuvettes (included with Colorimeter, or SE-8739)	4+
		Graduated cylinder, 10-mL	1
		Beakers, 50-mL	5
		Test tubes, 15-cm x 2-cm	5
		Test tube rack	1
		Chromatography chambers with lids†	6
		Chromatography paper strips, 2-cm wide	6
		Sep-Pak® C18 chromatography cartridge	1
		Syringe‡ with 4-mm (outer diameter) Leur tip,	1
		10-mL or larger	
		Disposable pipet	1
		70 % Isopropyl alcohol	30 mL
		Small plastic plate	1
		Toothpicks, flat	3
		Metric ruler	1
		Permanent marker or grease pencil	1
		Mechanical pencil	1
		Rinse bottle filled with distilled water	1
		Zip-seal sandwich bag	1
		Lint- and scratch-free lens wipes	2+
		Paper towels	1 roll*
		Distilled water	60 mL
		Waste beaker	1
		For teacher preparation:	
		0.1 % Sodium chloride, NaCl	1 L
		2.0 % Sodium chloride, NaCl	1 L
		5 % Isopropyl alcohol solution	1 L
		25 % Isopropyl alcohol solution	1 L
		1 % Red #40 solution	100 mL
		0.5 % Blue #1 solution	100 mL
		Unsweetened dilute Grape Kool-Aid <sup>TM</sup> solution	50 mL
		Unsweetened concentrated Grape Kool-Aid™	400 mL
		solution	
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†Use either commercial chambers, or construct each with a 400-mL beaker, paper clip, and aluminum foil

‡The syringe included with either the Wireless Pressure sensor (PS-3203) or the Wireless Drop Counter (PS-3214) is suitable for this investigation

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Lab #	Title	Materials and Equipment	Qty
6	A Chemistry Mystery: Name That Unknown!	For each student group: Computer or mobile device PASCO Wireless Conductivity sensor (PS-3210) PASCO Wireless pH sensor (PS-3204) Hot plate (PS-3401) Beakers, 100-mL Beaker, 250-mL Graduated cylinders, 10-mL Graduated cylinder, 100-mL Test tubes, 15-cm x 2-cm Test tube rack Stirring rod Tongs Table salt Table sugar Paraffin wax Aluminum foil squares, 5-cm x 5-cm Masking tape Permanent marker Rinse bottle filled with distilled water For teacher preparation: 0.10 M Sodium hydroxide, NaOH 0.10 M Hydrochloric acid, HCl Aspartame or Equal <sup>TM</sup> sweetener (C <sub>14</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> ), or xylitol (C <sub>3</sub> H <sub>12</sub> O <sub>3</sub> ) Potassium chloride, KCl Potassium bitartrate or cream of tartar (KC <sub>4</sub> H <sub>5</sub> O <sub>6</sub> ),	1 1 1 1 1 3 1 1 1 1 1 1 1 1 1 1 1 1 1
		Sodium carbonate, $Na_2CO_3$ Salicylic acid, $C_7H_6O_3$	<sup>1</sup> /4 cup <sup>1</sup> /4 cup
7	Stoichiometry in Solutions	For each student group:         Computer or mobile device         PASCO Wireless Conductivity sensor (PS-3210)         PASCO Wireless Temperature sensor (PS-3201)         PASCO Wireless Drop Counter with accessories         (PS-3214)         Beakers, 150-mL         Waste beaker, 250-mL         Graduated cylinder, 10-mL         Graduated cylinders or volumetric pipets with         bulbs, 25-mL         Ring stand         Magnetic stirrer with stir bar (SE-7700)         Phenolphthalein         Rinse bottle filled with distilled water         For teacher preparation:         1.0 M Sodium hydroxide, NaOH         0.10 M Hydrochloric acid, HCl	1 1 1 1 2 1 1 2 1 1 2 1 1 Dropper 1 1 L 1 L
		0.20 M Hydrochloric acid, HCl Hydrochloric acid, varied concentrations	1 L *

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Lab #	Title	Materials and Equipment	Qty
8	Percentage of H <sub>2</sub> O <sub>2</sub> In Your Drugstore	For each student group:	
	Hydrogen Peroxide	Computer or mobile device	1
		PASCO Oxidation Reduction Potential (ORP) Probe	1
		(PS-3515)†	
		PASCO Wireless Drop Counter with accessories	1
		(PS-3214)	
		Magnetic stirrer with stir bar (SE-7700)	1
		Precision balance (readability: 0.001 g; SE-8860)	1
		Ring stand	1
		Beakers, 150-mL	2
		Waste beaker, 250-mL	1
		Graduated cylinders, 10-mL	2
		Volumetric pipet with bulb, 10-mL	1
		Disposable pipet graduated to 2-mL	1
		$3.0$ % Hydrogen peroxide, $H_2O_2$	2 mL
		Distilled water	~250 mL
		Rinse bottle filled with distilled water	1
		For teacher preparation:	
		$4.0 M$ Sulfuric acid, $H_2SO_4$	500 mL
		$0.020 M Potassium permanganate, KMnO_4$	1 L
		0.10 M Ferrous ammonium sulfate (FAS),	1 L
		$Fe(NH_4)_2(SO_4)_2$	

†The ORP must be attached to PASCO's Wireless pH sensor box (PS-3204)

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Lab #	Title	Materials and Equipment	Qty
9	Investigating Physical and Chemical	For each student group:	
	Changes of Matter	Computer or mobile device	1
		PASCO Wireless pH sensor (PS-3204)	1
		PASCO Wireless Temperature sensor (PS-3201)	1
		PASCO Wireless Conductivity sensor (PS-3210)	1
		PASCO Wireless Pressure sensor with accessories	1
		Stopcock <sup>‡</sup> to fit syringe from Pressure sensor	1
		Graduated cylinder 50-mI	1
		Beakers 100-mI	6
		Stirring rod	1
		Test tubes 15-cm x 2-cm	2
		Two-hole stopper size #?	1
		Test tube rack	1
		Tongs	1
		Glycerin	Dropper
		Paper towels	*
		Labeling pen or wax pencil	1
		Rinse bottle filled with distilled water	1
		70 % Ethanol, $CH_3CH_2OH$	2 mL
		For teacher preparation:	
		Blue food coloring solution	1 L
		Yellow-orange food coloring solution	1 L
		0.10 M Hydrochloric acid, HCl	1 L
		0.10 M Sodium hydroxide, NaOH	1 L
		Steel wool soaked in vinegar	$\sim 8 \text{ cm}^3$
		Sucrose, $C_{12}H_{22}O_{11}$	Each
		Sodium chloride, NaCl	group
		Sodium acetate, NaCH <sub>3</sub> COO	requires a
		Calcium (Ca), metal turning	pea-sized
		Ammonium nitrate, NH <sub>4</sub> NO <sub>3</sub>	sample

+Stopcocks included with PASCO's Wireless Drop Counter (PS-3214) are suitable for this investigation

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Lab #	Title	Materials and Equipment	Qty
10	What Does Acid Rain Do to Coral Reefs?	For each student group:         Computer or mobile device         PASCO Wireless Pressure sensor with accessories         (PS-3203)         Erlenmeyer flasks, 250-mL         Graduated cylinder, 10-mL         Digital balance (readability: 0.01 g; SE-8823A)         Scoopula         Ring stand and flask clamp         Mortar and pestle         Calcium carbonate (CaCO <sub>3</sub> ) marble chips         Optional: Limestone samples         Optional: Blackboard chalk         Waste container         Rinse bottle filled with distilled water	1 1 1 1 1 1 1 25 g ** * * 1 1
		For teacher preparation: 1.0 M Hydrochloric acid, HCl 3.0 M Hydrochloric acid, HCl 6.0 M Hydrochloric acid, HCl	1 L 200 mL 200 mL 1 L
11A	Kinetics of Crystal Violet Fading (Colorimeter)	For each student group: Computer or mobile devicePASCO Wireless Colorimeter (PS-3215)Cuvettes (included with colorimeter, or SE-8739)Graduated cylinder, 10-mLTest tubes, 15-cm x 2-cmTest tube rackDisposable pipettesDistilled waterMarking penLint- and scratch-free lens wipesRinse bottle filled with distilled waterFor teacher preparation: 0.020 M Sodium hydroxide, NaOH 0.20 M Sodium hydroxide, NaOH0.20 M Sodium hydroxide, NaOH	1 1 1 1 4 1 5 3 mL 1 2+ 1 1 L 1 L 1 L

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Lab #	Title	Materials and Equipment	Qty
11B	Kinetics of Crystal Violet Fading (Spectrometer)	For each student group: Computer or mobile device with Spectrometry PASCO Wireless Spectrometer (PS-2600, or the	1 1
		Cuvettes† (included with spectrometer, or SE-8739) Graduated cylinder 10-mL	10
		Test tubes†, 15-cm x 2-cm	4†
		Test tube rack	1
		Disposable pipettes	5
		Distilled water Marking pen	3 mL
		Lint- and scratch-free lens wines	$\begin{vmatrix} 1\\2+ \end{vmatrix}$
		<i>Rinse bottle filled with distilled water</i>	1
		For teacher preparation:	1.7
		0.020 M Sodium hydroxide, NaOH	
		0.10 M Sodium hydroxide, NaOH	
		$2.5 \times 10^{-5} M$ Crystal violet solution	1 L
12	Building a Better Hand Warmer	For each student group.	
		Computer or mobile device	1
		PASCO Wireless Temperature sensor (PS-3201)	1
		Calorimeter <sup>‡</sup> (TD-8825A)	1
		Magnetic stirrer with stir bar (SE-7700)	1
		Graduated cylinder, 100-mL	
		Beaker, 250-mL Digital balance (readability: 0.01 g: SE 8823A)	
		Ring stand and ring clamp	1
		Heat-resistant gloves	1 pair
		Distilled water	500+ mL
		Waste container	1
		Magnesium sulfate, MgSO <sub>4</sub>	15 g
		Any combination of 3 of the following solids:	1.5
		• Ammonium chloride, $NH_4Cl$ • Calcium chloride, $CaCl$	15 g
		• Lithium chloride. LiCl	15 g
		• Sodium acetate, NaCH <sub>3</sub> COO	15 g
		• Sodium carbonate, Na <sub>2</sub> CO <sub>3</sub>	15 g
		• Sodium chloride, NaCl	15 g
		For teacher preparation:	
		Hot distilled water bath, ~80 °C	1

+1f 20.0-mL volumetric pipets with bulbs are not available, substitute with (2) 25-mL graduated cylinders plus (2) additional disposable pipettes

‡Or, substitute a calorimeter with two nested 10-oz polystyrene cups and a corrugated cardboard top

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Lab #	Title	Materials and Equipment	Qty
13	Applications of Le Châtelier's Principle	For each student group:Computer or mobile devicePASCO Wireless Colorimeter (PS-3215)Cuvettes (included with Colorimeter, or SE-8739)Beakers, 50-mLBeakers, 250-mLGraduated cylinder, 10-mLGraduated cylinder, 10-mLGraduated disposable pipettes, 3-mLTest tubes, 15-cm x 2-cmTest tube rackGlass stirring rodHot plate (PS-3401)IceCobalt(II) chloride, CoCl2ScoopulaMarking pen or grease pencilDistilled waterLint- and scratch-free lens wipesFor teacher preparation:0.0080 M Iron(III) nitrate, $Fe(NO_3)_3$ 0.0030 M Potassium thiocyanate, KSCN (to dilute to a 0.0010 M solution)Distribute in pipets or dropper bottles:	1 1 2 3 2 1 2 3 1 1 2 3 1 1 2 3 1 1 2 3 1 1 2 3 1 1 2 3 1 1 1 -125 mL 1.5 g 1 Dropper 2+ 1 L 1 L 1 L
		0.10 M Silver nitrate, AgNO <sub>3</sub> 6.0 M Hydrochloric acid, HCl	1 L 1 L
14	Investigation of Acid-Base Titrations	For each student group:Computer or mobile devicePASCO Wireless pH sensor (PS-3204)Wireless Drop Counter with accessories (PS-3214)Magnetic stirrer with stir bar (SE-7700)Beaker, 150-mLWaste beaker, 250-mLWaste beaker, 250-mLWaste beaker, 1-LGraduated cylinder, 50-mLDisposable pipetVolumetric pipets with bulbs†, 20-mLRing standPhenolphthalein solutionDistilled waterRinse bottle filled with distilled waterFor teacher preparation:0.10 M Sodium hydroxide, NaOH0.10 M Sodium hydroxide, NaOH0.10 M Sodium hydroxide, NaOH0.10 M Acetic acid, CH <sub>3</sub> COOH0.10 M Acetic acid, CH <sub>3</sub> COOH0.10 M Maleic acid, H2l	1 1 1 1 1 1 1 1 1 1 2 1 Dropper 200+ mL 1 1 L 1 L 1 L 1 L 1 L 1 L 1 L 1 L 1 L 1 L

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Lab #	Title	Materials and Equipment	Qty
15	Introduction to Buffers	For each student group: Computer or mobile device PASCO Wireless pH sensor (PS-3204) Beakers, 50-mL Beakers, 100-mL Graduated cylinder, 10-mL Graduated cylinders, 25-mL Graduated cylinder, 50-mL Stirring rod Spatula Mortar and pestle	1 1 5 2 1 2 1 1 1 1 1
		Disposable pipets Bufferin <sup>TM</sup> tablet, 325-mg Aspirin tablet, 325-mg Sodium acetate, NaCH <sub>3</sub> COO 5% Distilled white vinegar Distilled water Labeling pen or grease pencil Rinse bottle filled with distilled water Waste container	6 1 ~1 g 20 mL 120 mL 1 1 1
		<u>For teacher preparation</u> : All of the following are required to prepare additional solutions: 1.0 M Sodium hydroxide, NaOH 1.0 M Acetic acid, CH <sub>3</sub> COOH Sodium bisulfite, NaHSO <sub>3</sub> Sodium bicarbonate, NaHCO <sub>3</sub>	100 mL 100 mL 2.075 g 1.680 g
16	Evaluation of Lemonade as a Buffer	For each student group:Computer or mobile devicePASCO Wireless pH sensor (PS-3204)Wireless Drop Counter with accessories (PS-3214)Magnetic stirrer with stir bar (SE-7700)Beaker, 150-mLWaste beaker, 250-mLWaste beaker, 1-LGraduated cylinders, 25-mLDisposable pipetsRing standDistilled waterRinse bottle filled with distilled water	1 1 1 1 1 1 2 4 1 200+ mL 1
		For teacher preparation: $0.10 M$ Sodium hydroxide, NaOH $0.0080 M$ Citric acid, $H_3C_6H_5O_7$ Country Time® lemonade (made from powder mix)Variety of fruit-flavored powdered drink mixes(Kool-Aid®, Crystal Light®, Monster Energy®,Mountain Dew®, etc.)	2 L 500 mL ~235 mL/8 oz *

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# Activity by PASCO Sensors and Equipment

This table indicates which lab activity uses the sensors or special equipment listed.

Items available from PASCO	Lab activity where used
PASCO Wireless sensors	
PASCO Wireless Colorimeter (PS-3215) Includes 10 polystyrene cuvettes; Replacement cuvettes: SE-8739	1, 2A, 5, 11A, 13
PASCO Wireless Conductivity Sensor (PS-3210)	3, 6, 7, 9
PASCO Wireless Temperature Sensor (PS-3201)	7, 9, 12, 13
PASCO Wireless pH Sensor (PS-3204)	4, 6, 7, 8 (sensor box only), 9, 14, 15, 16
PASCO Oxidation-Reduction Potential Probe (PS-3515) Attaches to the Wireless pH sensor box (PS-3204)	8
PASCO Wireless Drop Counter (PS-3214) Accessories included: Syringe reservoir, drop tip, clamp, micro stir bar, 2 stopcocks	3, 5 (syringe only), 7, 8, 9 (stopcock only), 14, 16
PASCO Wireless Pressure Sensor (PS-3203) Accessories included: Syringe, tubing, connectors, size #6 one-hole rubber stopper	5 (syringe only), 9, 10
PASCO Wireless Spectrometer (VIS) (PS-2600) Includes 10 polystyrene cuvettes; Replacement cuvettes: SE-8739	2B, 11B
Lab equipment available from PASCO	
Precision balance readable to 0.001 g (SE-8860)	2A, 2B, 3, 8
Digital balance readable to 0.01 g (SE-8823A)	10, 12
Magnetic stirrer with stir bar (SE-7700)	3, 4, 7, 8, 12, 14, 16
Electrode support (PS-3505)	4
Hot plate (PS-3401)	6, 13
Calorimeter (TD-8825A)	12

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